

SHOCK TUBE EVALUATION OF HYDRAULIC FLUIDS

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INTRODUCTION

While the shock tube has been used extensively in the study of gas-phase reactions, relatively few studies have been made of gas-liquid reactions. The main reason for this has been the difficulty in interpreting the data obtained when two phases are initially present, because the additional steps of drop break-up, evaporation and mixing need to be considered, as well as chemical kinetics.

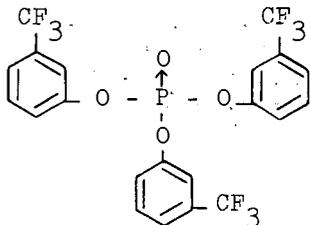
A study of the combustion process when n-hexadecane is injected as a fine spray into shock-heated air was made by Mullaney (Ref. 1). By high-speed photography he was able to observe injection of the spray, evaporation of the droplets, and spontaneous ignition. In most of his experiments combustion started before evaporation was complete. Shock tubes have also been used by Morrell and co-workers (Ref. 2,3) and by Hanson, Domich and Adams (Ref. 4) to study the break-up of liquid jets by rapidly moving gas streams. These investigators used inert liquids to study atomization of the liquids without the complications introduced by combustion, and were able to obtain equations relating atomization time to physical properties of the liquid and gas.

The original intent of this work was to develop a method of rating the ignition characteristics of fire-resistant hydraulic fluids that could be related to the single-cylinder engine test (Ref. 5) currently used, yet require much less fluid for an evaluation. The results shed some light on the parameters governing liquid-gas reactions.

EXPERIMENTAL

The shock tube used for these experiments has been described in detail elsewhere (Ref. 6). Briefly, it is 3 inches in diameter, with 12-foot low-pressure and 20-foot high-pressure sections. Shock speeds are measured by timing the passage of a shock wave between stations 55 and 7 inches from the closed end of the low-pressure section. Gas temperatures and flow velocities are calculated from the shock speeds. A piezoelectric pressure transducer is mounted in the top of the tube, 3 inches from the closed end, while a fused quartz window (covered except for a 2 mm vertical slit) is in the side of the tube, also 3 inches from the end. Light emitted by combustion in the tube is detected by a photo-multiplier tube 8 inches away from the tube, which also was covered except for a 2-mm vertical slit. Because of the slits, light emitted only from gas 3 inches from the end of the tube was detected.

Of a number of exploratory hydraulic fluids tested one, of formula



had the high ignition temperature of 1365°K., which from the extrapolation of Figure 2 would indicate an engine test rating of about 80.

DISCUSSION

There is, and probably will be for some time, a question as to the relative importance in the ignition process of the physical factors of drop break-up, evaporation, convective and diffusive mixing on the one hand, and chemical reactivity on the other.

Morrell and Povinelli (Ref. 3) have developed an equation for the time for break-up of liquid cylinders by shock waves, which should also apply approximately to drops. The break-up times of the 0.01 ml drops of "standard" liquids used in the above experiments have been calculated, as follows:

<u>Liquid</u>	<u>Calculated Break-up Time, milliseconds</u>
MS-2110-H	0.24
Fluid AV	0.31
Xylene	0.81
MIL-H-19457	0.25

These drop break-up times do not correlate with the ignition data, since the break-up times of the most and least flammable liquids are the same, while the calculated break-up times for Fluid AV and xylene, which have similar ignition temperatures, are different. Moreover, the boiling points of these latter two compounds differ considerably, being 325° and 140°C, respectively. It seems, therefore, that under these conditions the physical properties of the fluids are less important than chemical reactivity in controlling the ignition delays.

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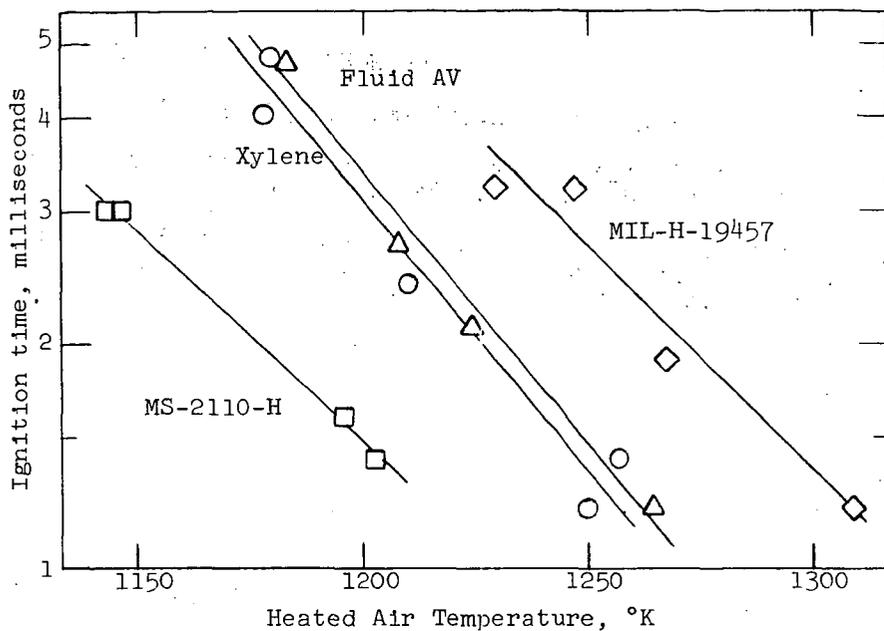


Figure 1. Shock tube ignition of hydraulic fluids

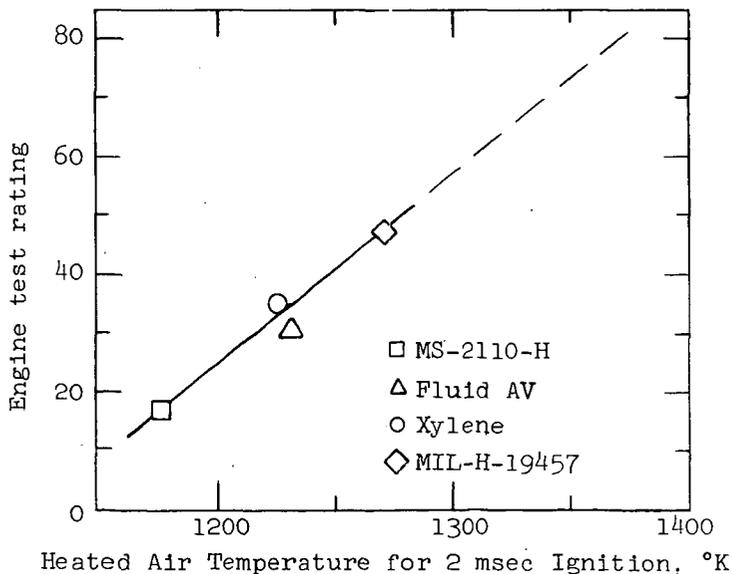


Figure 2. Calibration curve - shock tube versus engine test rating